

المعتقال سلامية العالمية مال فينتهن استبلاغ انتجارا بتجنيبا ملدين

PALM OLEIN-IN-WATER EMULSIONS STABILISED BY SPAN[®] AND TWEEN[®] SURFACTANTS AS POTENTIAL VEHICLES FOR DRUG DELIVERY

BY

MOHAMED SUFIAN BIN MOHD NAWI

A thesis submitted in fulfilment of the requirements for the degree of Master in Pharmaceutical Sciences (Pharmaceutical Technology)

> Kulliyyah of Pharmacy International Islamic University Malaysia

> > NOVEMBER 2008

ABSTRACT

Palm olein emulsions were produced using the combinations of Span[®] and Tween[®] surfactants by mechanical homogenisation. Effects of the types of surfactants, concentrations, effective HLB and the types of viscosity modifiers and concentrations on the characteristics of the emulsions were investigated. With palm olein content of 20% (w/w), stable oil droplets were produced at HLB values ranging from 8.5 to 11.0. Optimal concentrations of surfactants ranged from 25 to 30% (w/w to oil) depending on the types of the Span[®]/Tween[®] mixtures. Among the viscosity modifiers used, Carbopol[®]940 was the most effective. Suitable concentrations of Carbopol[®]940 for the emulsions prepared with Span[®]20/Tween[®]20 ranged from 0.1 to 0.3% (w/w). Beyond this concentration, destabilisation of emulsion due to at least depletion of water molecules could have occurred as a result of competitive hydration between Carbopol[®]940 and the surfactants. The emulsions produced exhibited viscoelastic and pseudoplastic behaviour, with vield value ranging from 0.1 to 35.2 Pa. Depending on the concentration of Carbopol[®]940 and within the linear viscoelastic region, the emulsions were elastic in nature as shown by domination of storage modulus (G') over the loss modulus (G") and with tan $\delta < 1$ in the frequency range of 0.01 to 10 Hz. These favourable rheological properties were induced by the formation of threedimensional network of Carbopol[®]940 molecules in the continuous aqueous phase, which also entrapped the oil droplets and thus increased the stability of the emulsion. To the optimised o/w emulsion formulation, active pharmaceutical ingredients and extracts of *Cassia alata* were incorporated. Betamethasone dipropionate and tolnaftate at concentrations of 0.5 mg/g and 1 mg/g respectively, did not affect the size of droplets and stability of the palm olein emulsions. Nonetheless, betamethasone dipropionate increased the viscosity and elasticity of the emulsion; while tolnaftate slightly reduced the viscosity, thixotropy and elasticity of the emulsion. Cassia alata extracts affected the emulsion stability. With up to 0.5 mg/g of ethanol extract and 0.25 mg/g of ethanol/water extract of *Cassia alata*, emulsions were stable. Further, Cassia alata extracts did not alter the pseudoplasticity of the emulsion despite decreasing viscosity with increasing concentration of the extracts. Desired rheological properties for the development of topical cream and lotion can be attained by changing the concentration of Carbopol[®]940 in palm olein-in-water emulsion, which is a potential vehicle for drug delivery.

ملخص البحث

تم انتاج مستحلبات زيت النخيل باستخدام الخليط من سبان ("Span) وتوين (Tween[®]) سرفقتنات (surfactants) بطريقة التجانس الميكانيكي (mechanical homogenisation). كما تم التحقق من آثار أنواع سرفقتنات, والتركيزات، وفعّال HLB وأنواع من معدلات اللزوجة، بإضافة إلى التركيز إت على خصائص المستحلبات. أنتجت قطر إت النفط المستقرة بنسبة 20 ٪ (وزن / وزن) من محتوى زيت النخيل فيما يتراوح بين 8,5 إلى 11,0من قيم معدلات HLB. والتركيزات المثلى لشرفقتنات ما يتراوح من 25 إلى 30 ٪ على حسب نوعية المخاليط من سِبَنْ (Span[®]) وتُوين (Tween[®]). كانCarbopol[®]940 هو الأكثر فعالة من بين معدلات اللزوجة المستخدمة. كما تم إعداد التركيزات المناسبة من Carbopol[®]940 للمستحلبات ب وفرق هذه 0,1 و0,1 الذي يتراوح بين 0,1 إلى 0,3 (وزن / وزن). وفوق هذه Span 20^{\degree} التركيزات، يمكن أن يحدث عدم استقرار المستحلب على الأقل، وهذا بسبب نضوب جزئيات المياه، نتيجة تنافس الهيدراسن (hydration) بين Carbopol[®]940 وسرفقنات. وتتصف المستحلبات المنتجة لزجا مطاطيا (viscoelastic) و pseudoplastic و yield value فيما تتراوح من 1.0 إلى 35,2 على حسب تركيزات 640°Carbopol، وداخل طول منطقة لزج مطاطى (linear viscoelastic region) وكانت المستحلبات مرونة بطبيعة كما اتضحت من سيطرة معامل التخزين (storage modulus) على معامل الفقد (loss modulus) و بطريقة $(\tan \delta < 1)$ وهذه الخصائص الطبيعية المواتية (tan $\delta < 1$. وهذه الخصائص الطبيعية المواتية كانت ناجمة عن تشكيل شبكة ثلاثية الأبعاد من جزئيات Carbopol[®]940 في استمر ار المرحلة المائية، التي تحاصر قطرات النفط وبالتالي تزيد في استقرار المستحلب. بصياغة المستحلب بالأداء الأمثل، وخلطه بالمكونات الصيدلية الناشطة والمقتطفات من Cassia alata. ولم يؤثر كل من بيتاميثاسون دبروبيونات dipropionate) (betamethasone وتُلنَفتات (tolnaftate) في تركيزات g/mg 0،5 في حجم قطرات واستقرار مستحلب زيت النخيل. ومع ذلك بيتاميثاسون دبر وبيونات تزيد في لزوجة ومرونة المستحلب بينما ثلنفتات تخفّض قليلا من لزوجة وthixotropy ومرونة المستحلب. ولا تؤثر مقتطفات Cassia alata في استقرار المستحلب الا أن يصل إلى ما بعد g/mg 0,5 من الإيثانول وما فوق g/mg 0,25 لإيثانول بالماء المستخرج من Cassia alata. علاوة على ذلك، وإن مقتطفات Cassia alata لم تغير pseudoplasticity المستحلب على الرغم من تناقص اللزوجة مع التركيز المتزايد للمقتطفات. ويمكن أن تتحقق الخصائص الطبيعية المرجوة لتطوير وتنمية كريم ومستحضر عن طريق تغيير تركيز Carbopol[®]940 في مياه مستحلب زيت النخيل التي هي الوسيلة الممكنة لإيصال الأدوية

APPROVAL PAGE

I certify that I have supervised and read this study and that in my opinion; it conforms to acceptable standards of scholarly presentation and is fully adequate, in scope and quality, as a thesis for the degree of Master in Pharmaceutical Sciences (Pharmaceutical Technology).

Kausar binti Ahmad Supervisor

I certify that I have read this study and that in my opinion it conforms to acceptable standards of scholarly presentation and is fully adequate, in scope and quality, as a thesis for the degree of Master in Pharmaceutical Sciences (Pharmaceutical Technology).

Maryanto Examiner

I certify that I have read this study and that in my opinion it conforms to acceptable standards of scholarly presentation and is fully adequate, in scope and quality, as a thesis for the degree of Master in Pharmaceutical Sciences (Pharmaceutical Technology).

Esti Hendradi External Examiner

This thesis was submitted to the Department of Pharmaceutical Technology and was accepted as a fulfilment of the requirements for the degree of Master in Pharmaceutical Sciences (Pharmaceutical Technology).

Kausar binti Ahmad Head, Department of Pharmaceutical Technology This thesis was submitted to the Kulliyyah of Pharmacy and was accepted as a fulfilment of the requirements for the degree of Master in Pharmaceutical Sciences (Pharmaceutical Technology).

Tariq bin Abdul Razak Dean, Kulliyyah of of Pharmacy

DECLARATION

I hereby declare that this thesis is the result of my own investigations, except where otherwise stated. I also declare that it has not been previously or concurrently submitted as a whole for any other degrees at IIUM or other institutions.

Mohamed Sufian Bin Mohd Nawi

Signature.....

Date.....

INTERNATIONAL ISLAMIC UNIVERSITY MALAYSIA

DECLARATION OF COPYRIGHT AND AFFIRMATION OF FAIR USE OF UNPUBLISHED RESEARCH

Copyright ©2008 by Mohamed Sufian Bin Mohd Nawi. All rights reserved.

PALM OLEIN-IN-WATER EMULSIONS STABILISED BY SPAN[®] AND TWEEN[®] SURFACTANTS AS POTENTIAL VEHICLES FOR DRUG DELIVERY

No part of this unpublished research may be reproduced, stored in a retrieval system, or transmitted, in any form or by any means, electronic, mechanical, photocopying, recording, or otherwise without the prior written permission of the copyright holder except as provided below.

- 1. Any material contained in or derived from this unpublished research may only be used by others in their writing with due acknowledgement.
- 2. IIUM or its library will have the right to make and transmit copies (print or electronic) for institutional and academic purposes.
- 3. The IIUM library will have the right to make, store in a retrieval system and supply copies of this unpublished research if requested by other universities and research libraries.

Affirmed by Mohamed Sufian Bin Mohd Nawi

Signature

Date

ACKNOWLEDGEMENTS

All praise is due to Allah S.W.T because of His bounty I can complete this research towards fulfilling the requirements for my master degree.

I would like to express my most sincere gratitude to my supervisor, Assistant Professor Dr. Kausar Ahmad for her help and guidance, and for her endless supply of suggestions and encouragement throughout this work at the Kulliyyah of Pharmacy, International Islamic University Malaysia, Kuantan.

I would like to express my utmost gratitude to my wife, Siti Hadijah and my children: Afifah, Amirul and Amirah for their moral support that gave me the confidence, will and strength to endure pressure and tension in pursuing my ambition and to them I dedicate this thesis.

My thanks to Dr. Tanveer, Dr. Norazian and Dr. Muhammad Taher for their assistance. Thanks to all Lecturers, Science Officers, Assistance Science Officers, Laboratory Technicians and administrative staff of Kulliyyah of Pharmacy and Kulliyyah of Medicine, IIUM for their kind assistance and support. The help, motivation and friendship from postgraduate friends will be cherished.

Finally but not least, thank you to my parents and mother-in-law for their prayers, understanding and inspiration throughout my study. May Allah gives His blessings in their lives in this world and in the hereafter.

TABLE OF CONTENT

Abstract	i
Abstract in Arabic	
Approval Page	
Declaration	
Declaration of Copyright and Affirmation of Fair Use of Unpublished Research	
Acknowledgements	
List of Tables	
List of Figures	
List of Abbreviations	
List of Symbols	
CHAPTER 1: INTRODUCTION	1
1.2 Use Of Vegetable Oils In Pharmaceuticals	
1.2.1 Palm oil as an alternative	
1.2.1.1 Constituents of palm oil	
1.3 Role Of Surfactants In Oil-In-Water Emulsions	
1.3.1 Factors affecting the properties and stability of emulsion	
1.3.2 Natural surfactants in palm olein	
1.3.3 Use of non-ionic surfactants	
1.3.3.1 HLB consideration.	
1.4 Rheology	
1.5 Thickeners In Pharmaceuticals	
1.5.1 Type of viscosity inducing agent	
1.6. Characterisation Of Emulsion	
1.7 Aim Of The Study	
CHAPTER 2: PREPARATION OF PALM OLEIN-IN-WATER EMULS USING SPAN®/TWEEN® SURFACTANTS	
2.1 Introduction	
2.2 Materials And Methods	20
2.2.1. Oil	20
2.2.2 Surfactants	
2.2.3 Experimental flow	
2.2.4 Preparation of emulsions	
2.2.4.1 Preparation of emulsions for determination of the opt	
ratio of mixed surfactants	
2.2.4.2 Preparation of emulsions to determine the effects of surf	
concentration	
2.2.4.3 Preparation of emulsions to determine the effect of var	
in HLB at optimum surfactant concentration	
2.2.4.4 Preparation of emulsions to determine opt	
homogenisation period for the purpose of process scale-up	
different types of surfactant mixtures	
2.2.5 Characterization	

2.2.5.1 Particle size determination	28
2.2.5.2. Phase separation	
2.2.5.3 Determination of rheological properties	29
2.2.5.4 Analysis of data	29
2.3 Results And Discussion	29
2.3.1 Optimum ratio of Span [®] /Tween [®] mixtures	29
2.3.2 Effects of surfactant concentration	31
2.3.2.1 Effect of surfactant concentration on initial droplet size	31
2.3.2.2 Effect of surfactant concentration on creaming behaviou	ır and
viscosity	36
2.3.3 Effect of HLB at optimum surfactant concentration	42
2.3.3.1 Effect of HLB on initial droplet size	42
2.3.3.2 Effect of HLB on creaming behaviour and viscosity	48
2.3.4 The effects of type of surfactant and quantity of emulsion of	on the
duration of homogenisation	52
2.4. Conclusions	54

EMULSIONS	55
3.1 Introductions	55
3.1.1 Bentonite	56
3.1.2 Tragacanth gum	57
3.1.3 Sodium carboxymethyl cellulose (CMC)	59
3.1.4 Carbomer (Carbopol [®])	60
3.1.4.1 Carbopol [®] 940	62
3.2 Materials And Methods	62
3.2.1 Viscosity inducing-agents	
3.2.2 Preparation of 1% (w/w) bentonite stock dispersion	
3.2.3 Preparation of 1% (w/w) tragacanth gum stock solution	
3.2.4 Preparation of 1% (w/w) carboxymethylcellulose sodium	(CMC)
stock solution	
3.2.5 Preparation of 1% (w/w) Carbopol [®] 940 stock solution	64
3.2.6 Preparation of stock emulsions	
3.2.7 Preparation of emulsions with viscosity inducing-agents	
3.2.8 Characterisation	
3.2.8.1 Phase separation	
3.2.8.2 Droplet size determination	
3.2.8.3 Determination of rheological properties	
3.3 Results And Discussion	
3.3.1 Effect of viscosity inducing-agents on rheological properties.	
3.3.3 Effect of viscosity inducing-agents on droplet size	
3.4 Conclusions	80
CHAPTER 4: THE EFFECT OF CARBOPOL [®] 940 CONCENTRATION	
PALM OLEIN-IN-WATER EMULSIONS	
4.1 Introduction	

4.1 Introducti	on		 	 	. 82
4.2 Materials	And Metho	ds	 	 	. 83
				concentration	

Carbopol [®] 940	
4.2.3 Commercial products	
4.2.4 Characterisation	
4.2.4.1 Determination of particle size and creaming behaviou	r 84
4.2.4.2 Determination of rheological properties	
4.2.4.2.1 Measurement using Brookfield D	
Programmable Viscometer	
4.2.4.2.2 Measurement using Haake MARS rheometer	
4.3 Results And Discussion	
4.3.1. Effect of Carbopol [®] 940 concentration on creaming behavio	ur 86
4.3.2 Effect of Carbopol [®] 940 concentration on droplet size	
4.3.3 Effects of Carbopol [®] 940 on rheological properties	
4.3.3.1 Effects of Carbopol [®] 940 on viscosity	
4.3.3.2 Effects of Carbopol [®] 940 on yield stress	
4.3.3.3.1 Viscosity, yield stress and thixotropy	
4.3.3.3.2 Viscoelastic properties	
4. 4 Conclusions	
CHAPTER 5: PALM OLEIN-IN-WATER EMULSION AS VEHICI	LE FOR
DRUG DELIVERY	105
5.1 Introduction	
5.1.1 Cassia alata	
5.1.2 Betamethasone dipropionate	
5.1.3 Tolnaftate	
5.2 Materials And Methods	
5.2.1 Materials	
5.2.2 Preparation of emulsions containing extracts of C.alata	109
5.2.2.1 Extraction Procedure	
5.2.2.2 Preparation of base emulsion	
5.2.2.3 Preparation of C. alata extract solutions	
5.2.2.4 Preparation of emulsions with various concentration	
alata extracts	
5.2.3 Preparation of betamethasone dipropionate emulsion	
5.2.3.1 Preparation of stock emulsion	
5.2.3.2 Preparation of 1% (w/w) Carbopol [®] 940 stock solution	
5.2.3.3 Preparation of betamethasone dipropionate slurry	
5.2.3.4 Preparation of emulsion with betamethasone dip	
equivalent to 0.5 mg/g betamethasone	
5.2.4 Preparation of 1 mg/g tolnaftate emulsion	
5.2.5 Characterisation	
5.2.5.1 Phase separation	
5.2.5.2 Particle size determination	114
5.2.5.3 Determination of rheological properties	
5.3 Results And Discussion	
5.3.1 Effect of C. alata extracts on the emulsion properties and	
5.5.1 Effect of C. alata extracts of the emulsion biodeffies and	115
	115 stability
	115 stability 115
5.3.1.1 Effect of C. alata extracts on creaming behaviour	115 stability 115 115
	115 stability 115 115 118

5.3.2 Effect of betamethasone dipropionate and tolnaftate on the emulsion
properties and stability
5.3.2.1 Effect of betamethasone dipropionate and tolnaftate on the
creaming behaviour
5.3.2.2 Effect of the betamethasone dipropionate and tolnaftate on the
size of droplet124
5.3.2.3 Effect of the betamethasone dipropionate and tolnaftate on the
rheological properties126
5.3.2.3.1 Viscoelastic properties of emulsions with
betamethasone dipropionate and tolnaftate
5.4 Conclusion
BIBLIOGRAPHY
APPENDIX A: STATISTICAL ANALYSIS: EFFECT OF
SPAN [®] 20/TWEEN [®] 20 CONCENTRATION ON SIZE AND POLYDISPERSITY
OF OIL DROPLET (RESULTS ILLUSTRATED IN FIGURE 2.4)145
OF OIL DROPLET (RESULTS ILLUSTRATED IN FIGURE 2.4)145
OF OIL DROPLET (RESULTS ILLUSTRATED IN FIGURE 2.4) 145 APPENDIX B: LIST OF WORKS PRESENTED
APPENDIX B: LIST OF WORKS PRESENTED 164
APPENDIX B: LIST OF WORKS PRESENTED

LIST OF TABLES

Table No.		Page No.
1.1	Characteristics of palm oil and its fractions.	6
2.1	Characteristics of palm olein.	21
2.2	Emulsion formulations for the determination of optimum	26
	concentration of Span [®] /Tween [®] .	
2.3	Emulsion formulations for the determination of optimum	27
	HLB of mixed surfactants.	
2.4	Emulsion formulations to determine the effect of the types of	27
	mixed surfactants and quantity of emulsions on optimum	
	homogenisation duration.	
2.5	Particle size distribution of emulsions.	35
2.6	Particle size distribution of emulsions at optimum range of	47
	HLB of mixed surfactants.	
3.1	Formulations for the preparation of stock emulsions.	64
3.2	Formulation of the palm olein emulsions with various type	78
	and concentration of viscosity inducing-agents, particle sizes	
	and yield stresses. The emulsions emulsified with	
	$\text{Span}^{\text{\$}}20/\text{Tween}^{\text{\$}}20$ (values are mean \pm SD, n = 3).	
3.3	Formulation of the palm olein emulsions with various type	79
	and concentration of viscosity inducing-agents, particle sizes	
	and yield stresses. The emulsions emulsified with	
	$\text{Span}^{\text{\$}}20/\text{Tween}^{\text{\$}}80$ (values are mean \pm SD, n = 3).	
4.1	Formulations for the preparation of stock emulsions.	83
4.2	Formulations and yield stress values of the emulsions with	97
	varied concentrations of Carbopol [®] 940.	
4.3	Rheological properties of palm olein emulsions containing	100
	varied concentration of Carbopol [®] 940 and commercial	
	emulsions (values are mean \pm SD, n = 3).	
5.1	Type and content of C. alata extracts in the emulsions.	112
5.2	Rheological properties of emulsion containing	128
	betamethasone dipropionate (BET) and tolnaftate (TL)	
	(values are mean \pm SD, n = 3).	

LIST OF FIGURES

2.1 Structures of (a) sorbitan monoesters (Span [®]) and (b)	22
polyoxyethylene 20 sorbitan monoester (Tween [®]).	
2.2 (a) Ultra-Turrax [®] T25 Basic homogeniser equipped wi	ith 25
dispersing tool S25N-25GM and (b) dispersing tool S2	25N-
25GM head.	
2.3 Size of droplets immediately after emulsification.	30
Emulsions prepared with 20% (w/w) palm olein and 10	
(w/w) of the mixed of surfactants of Span [®] 20/Tween [®] .	20
and Span [®] 20/Tween [®] 80.	
2.4 Size of droplets immediately after emulsification.	30
Emulsions prepared with 20% (w/w) palm olein and 10	
(w/w) of the mixture of surfactants of Span [®] 40/Tween	[®] 20
and Span [®] 40/Tween [®] 80.	
2.5 Effect of Span [®] 20/Tween [®] 20 concentration on size and	d 32
polydispersity of oil droplet (values are mean \pm SD, n	
2.6 Effect of Span [®] 20/Tween [®] 80 concentration on size and	
polydispersity of oil droplet (values are mean \pm SD, n	
2.7 Effect of Span [®] 40 /Tween [®] 20 concentration on size and	
polydispersity of oil droplet (values are mean \pm SD, n	
2.8 Effect of Span [®] 40 /Tween [®] 80 concentration on size and	
polydispersity of oil droplet (values are mean \pm SD, n	= 3).
2.9 Creaming behaviour and apparent viscosities at 1200 s	s-1 as 37
a function of concentration of mixed surfactants.	
2.10 Viscosities of 20% palm olein emulsions measured at a	a 38
shear rate of 1200s-1 as a function of concentration of	
mixed surfactants.	
2.11 Phase separation after two months and initial particle s	
of 20% palm olein emulsion with Span [®] 20/Tween [®] 20	at
various HLB (values are mean \pm SD, n = 3).	
2.12 Phase separation after two months and initial particle s	
of 20% palm olein emulsion with Span [®] 20/Tween [®] 80	at
various HLB (values are mean \pm SD, n = 3).	
2.13 Phase separation after two months and initial particle s	
of 20% palm olein emulsion with Span [®] 40/Tween [®] 20	at
various HLB (values are mean \pm SD, n = 3).	

2.14	Phase separation after two months and initial particle size of 20% palm olein emulsion with Span [®] 40/Tween [®] 80 at	44
	various HLB (values are mean \pm SD, n = 3).	
2.15	Viscosities of 20% palm olein emulsions measured at a	50
	shear rate of 1200s-1 as a function of HLB of mixed	
	surfactants.	
2.16	Effect of duration of homogenisation on droplet size of	52
	20% (w/w) of palm olein emulsified by	
	Span [®] 20/Tween [®] 20 and Span [®] 40/Tween [®] 20 at their	
	respective optimum concentration and optimum HLB (refer	
	to Table 2.4). The quantities of emulsions prepared were	
	(a) 150g and (b) 1700g.	
3.1	The main chemical building blocks of the carbohydrate	58
	biopolymer tragacanthh gum. (a) β -D-xylose, (b) L-	
	arabinose, (c) α -D-galacturonic acid, (d) α -D-galacturonic	
	acid methylester and (e) β -Dgalactose, (f) α -L-fucose.	
	(Sourced from Mohamadnia et al., 2008).	
3.2	Structure of sodium carboxymethyl cellulose. (Sourced	59
	from Guo, et al., 1998)	
3.3	Schematic representation of the structure of Carbomer	61
	(Carbopol [®]) in acid (top) and neutral pH (bottom). Sourced	
	from Gallardo, Ruiz and Delgado (2000).	
3.4	Viscosity curves of 18% palm olein emulsions with various	68
	concentration of bentonite. The emulsions were prepared	
	with (a) $\text{Span}^{\$}20/\text{Tween}^{\$}20$ and (b) $\text{Span}^{\$}20/\text{Tween}^{\$}80$.	
3.5	Viscosity curves of 18% palm olein emulsions with various	68
	concentration of tragacanth gum. The emulsions were	
	prepared with (a) Span [®] 20/Tween [®] 20 and (b)	
	Span [®] 20/Tween [®] 80.	
3.6	Viscosity curves of 18% palm olein emulsions with various	69
	concentration of CMC. The emulsions were prepared with	
	(a) Span [®] 20/Tween [®] 20 and (b) Span [®] 20/Tween [®] 80.	
3.7	Viscosity curves of 18% palm olein emulsions with various	69
	concentration of Carbopol [®] 940. The emulsions were	
	prepared with (a) $\text{Span}^{\mathbb{R}}20/\text{Tween}^{\mathbb{R}}20$ and (b)	
	Span [®] 20/Tween [®] 80.	
3.8	Viscosity curves of 18% palm olein emulsions with 0.1%	70
	bentonite, tragacanth gum, CMC and Carbopol [®] 940. The	

	emulsions were prepared with (a) Span [®] 20/Tween [®] 20 and (b) Span [®] 20/Tween [®] 80.	
3.9	Creaming behaviour after two months of 18% palm olein	72
	emulsions as a function of concentration of bentonite.	
	Emulsions were stabilised by (a) Span [®] 20/Tween [®] 20 and	
	(b) $\text{Span}^{\&}20/\text{Tween}^{\&}80$.	
3.10	Creaming behaviour after two months of 18% palm olein	73
	emulsions as a function of concentration of tragacant gum.	
	Emulsions were stabilised by (a) Span [®] 20/Tween [®] 20 and	
	(b) Span [®] 20/Tween [®] 80.	
3.11	Creaming behaviour after two months of 18% palm olein	73
	emulsions as a function of concentration of CMC.	
	Emulsions were stabilised by (a) Span [®] 20/Tween [®] 20 and	
	(b) Span [®] 20/Tween [®] 80.	
3.12	Creaming behaviour after two months of 18% palm olein	74
	emulsions as a function of concentration of Carbopol [®] 940.	
	Emulsions were stabilised by (a) Span [®] 20/Tween [®] 20 and	
	(b) Span [®] 20/Tween [®] 80.	
4.1	Phase separation of 20% palm olein emulsion stabilised by	87
	Span [®] 20/Tween [®] 20 at various concentration of	
	Carbopol [®] 940 (values are mean \pm SD, n = 3).	
4.2	Phase separation of 20% palm olein emulsion with	88
	Span [®] 20/Tween [®] 80 at various concentration of	
	Carbopol [®] 940 (values are mean \pm SD, n = 3).	
4.3	Initial droplet size and droplet size after 6 months $(\bullet, \blacktriangle)$	90
	and 11 months (\Box, Δ) as a function of the concentration of	
	Carbopol [®] 940 in emulsions prepared with 20% (w/w) of	
	palm olein and 25% (w/w) to oil of Span [®] 20/Tween [®] 20	
	(values are mean \pm SD, n = 3).	
4.4	Initial droplet size and droplet size after 6 months as a	91
	function of the concentration of Carbopol [®] 940 in	
	emulsions prepared with 20% (w/w) of palm olein and 30%	
	(w/w) to oil of Span [®] 20/Tween [®] 80 (values are mean \pm SD,	
	n = 3).	
4.5	Viscosity of 20% (w/w) palm olein emulsions with 0.1%	93
	(w/w) of Carbopol [®] 940 as a function of shear rate. The	
	emulsions were stabilized with Span [®] 20/Tween [®] 20 (Δ) and	
	Span [®] 20/Tween [®] 80 (■).	

4.6	Viscosity of commercial topical creams (EC and EV) and	93
	emulsions containing 20% (w/w) of palm olein with \mathbb{R}^{20}	
	Span [®] 20/Tween [®] 20 and varied concentration of C_{1} = 1 [®] 0.40 m f = 0.1 m f = 0	
4.7	Carbopol [®] 940 as a function of shear rate.	0.4
4.7	Viscosity of commercial topical creams (EC and EV) and	94
	emulsions containing 20% (w/w) of palm olein with $R = \frac{R}{20} \sqrt{T}$	
	$\text{Span}^{\mathbb{R}}20/\text{Tween}^{\mathbb{R}}80$ and varied concentration of	
4.0	Carbopol [®] 940 as a function of shear rate.	0.6
4.8	The effect of the concentration of Carbopol [®] 940 on the $\frac{1}{2}$	96
	yield stress of emulsions containing 20% (w/w) palm olein	
	with Span [®] 20/Tween [®] 20 (*) and Span [®] 20/Tween [®] 80 (\blacklozenge).	
4.9	Flow curves (\times) and viscosity curves (\bullet) after 6 months of	99
	the emulsions with (a) 0.1% (w/w), (b) 0.2% (w/w) and (c)	
	0.3% (w/w) of Carbopol [®] 940. The emulsions consisting of	
	20% palm olein and 25% of Span [®] 20/Tween [®] 20.	
4.10	Storage modulus (G'), loss modulus (G'') and tan δ of palm	101
	olein emulsions with varied concentration of Carbopol [®] 940	
	as a function of frequency (f).	
4.11	Complex modulus (G^*) of EV cream (\times), EU body lotion	102
	(\checkmark) , SE oral emulsion (+) and palm olein emulsions with	
	0.1% ($^{\circ}$), 0.2% ($^{\Delta}$) and 0.3% ($^{\Box}$) of Carbopol [®] 940 as a	
	function of frequency (f).	
5.1	Phase separation (after 1 hour, 1 day, 2 weeks and 8	117
	months) of 20% palm olein emulsions as a function of	
	concentration of C. alata: (a) water extract, (b) ethanol	
	extract and (c) ethanol/water extract. The emulsions were	
	stabilised by Span [®] 20/Tween [®] 20 and thickened with 0.3%	
	(w/w) of Carbopol [®] 940.	
5.2	Droplet size (after two weeks) as a function of the	118
	concentration of C. alata water extract (CAW), ethanol	
	extract (CAE) and ethanol/water extract (CAEW) (values	
	are mean \pm SD, n = 3).	110
5.3	Droplet size (after 8 month) as a function of the	119
	concentration of C. alata water extract (CAW), ethanol	
	extract (CAE) and ethanol/water extract (CAEW) (values	
. .	are mean \pm SD, n = 3).	
5.4	Viscosity curve of the emulsions with varied concentration	122

	of C. alata: (a) water extract, (b) ethanol extract and (c) ethanol/water extract.	
5.5	Phase separation (after 1 day, 2 weeks, 1 month and 3 months) of 20% palm olein emulsions with betamethasone dipropionate and tolnaftate. The emulsions were stabilised by Span [®] 20/Tween [®] 20 and thickened with 0.2% (w/w) of Carbopol [®] 940.	124
5.6	Droplet size as a function of storage period of the emulsion with (a) betamethasone dipropionate (BET) and (b) tolnaftate (TL) (values are mean \pm SD, n = 3).	125
5.7	Flow curves (×) and viscosity curves (•) of: (a) control emulsion at pH 5.7, (b) betamethasone emulsion (pH = 5.7), (c) control emulsion at pH 6.5 and (d) tolnaftate emulsion (pH = 6.5).	126
5.8	Storage modulus (G'), loss modulus (G'') and tan δ of emulsion with betamethasone dipropionate (BET) and control as a function of oscillation shear stress.	129
5.9	Storage modulus (G'), loss modulus (G'') and tan δ of emulsion with tolnaftate (TL) and control as a function of oscillation shear stress.	130
5.10	Storage modulus (G'), loss modulus (G'') and tan δ of emulsion with betamethasone dipropionate (BET) and control as a function of frequency.	131
5.11	Storage modulus (G'), loss modulus (G'') and tan δ of emulsion with tolnaftate (TL) and control as a function of frequency.	132

LIST OF ABBREVIATIONS

A _{cmc}	area per molecule just below the cmc
API	active pharmaceutical ingredients
BET	betamethasone dipropionate
BNF	British National Formulary
BP	British Pharmacopoeia
CAE	ethanol extract of C. alata
CAEW	ethanol/water extract of C. alata
CAW	water extract of C. alata
CMC	sodium carboxymethyl cellulose
cmc	critical micelle concentration
CR	controlled rate
D(v,0.1)	diameter at which 10% of the volume falls below it
- ()	
D(v,0.5)	diameter at which 50% of volume of particles falls below it or
	median diameter
D(v,0.9)	diameter at which 90% of volume of particles falls below it
D[4,3]	volume average diameter
EO	oxyethylene
HLB	hydrophile-lipophile balance
LVR	linear viscoelastic region
MIMS	Malaysian Index of Medical Specialities
PPG	polypropylene glycol
RO	reverse osmosis
SD	standard deviation
tan δ	tangent of the phase angle
TL	tolnaftate
USP/NF	United State Pharmacopoeia/National Formulary

LIST OF SYMBOLS

τ	shear stress
$\dot{\gamma}$	shear rate
au°	yield stress
v	terminal velocity
f	frequency
G*	complex modulus
G'	storage modulus
G"	loss modulus
r	radius of the particle
δ	phase angle
η	viscosity

CHAPTER 1

INTRODUCTION

An emulsion is a dispersed system containing at least two immiscible liquid phases. Emulsions can be divided into either oil-in-water (o/w) or water-in-oil (w/o) emulsions. In the o/w emulsion oil droplets are dispersed throughout the aqueous phase, while in w/o emulsion aqueous is dispersed throughout the oil phase. In addition to these simple emulsions, multiple emulsions can also be prepared. These can be either o/w/o or w/o/w emulsions.

Generally, emulsion is one of the most widely used dosage form in pharmaceutical delivery system and is an attractive vehicle for the administration of poorly soluble drugs. Pharmaceutical and cosmetic form of emulsions can be prepared in the form of liquids or semisolids, based on intended application. Pharmaceutically, the roots of administration of emulsion are oral, topical, mucosal, parenteral and ophthalmic.

There are many advantages and interest in lipid-emulsions formulations. The advantage of oral emulsion preparation is its ability to prepare formulation for easy compliance especially for paediatric and geriatric patients (Marti-Mestres and Nielloud, 2000). For total parenteral nutrition, fat emulsion prepared from vegetable oil emulsified by phospholipids (eg. Intralipid[®]) is not recognised as a foreign substance by the body (Buszello and Muller, 2000). Topical emulsion is less greasy and more aesthetically appealing to patients (Thomson, 2004). For skin irritant medicinal agents, incorporation into the emulsion reduces irritation (Howard, Loyd and Nicholas, 1999). In ophthalmic emulsion formulation, it gives an advantage in

providing a vehicle for slow release of drug in a nonviscous medium, thus reducing the frequency of application (Fialho and Silva-Cunha, 2004).

There is a possibility of dissolving lipophilic components such as propofol in a safe and tolerable matrix of emulsion with the advantage of having less pain upon injection compared with solvent based. Lipid emulsions also minimise losses of more lipophilic drugs during infusion into plastic tubing and infusion sets (Collins-Gold, Feichtinger and Warnhein, 2000).

1.1 USE OF MINERAL OIL

Mineral oil is a mixture of aliphatic hydrocarbons obtained from petroleum. They are available as liquid hydrocarbon (light liquid paraffin), solid hydrocarbon (hard paraffin) and semisolid hydrocarbon (white soft paraffin) (Reilly, 2000) and are widely used as excipients in cosmetics and pharmaceuticals. Most of the topical cosmetic and pharmaceutical and oral emulsion preparations are mineral oil based such as betamethasone cream (Betnovate® cream) using white soft paraffin and phenolphthalein (Agarol®) using liquid paraffin and others (Walker, 1991-92).

Unfortunately, the source of fossil oil is diminishing and the price of petroleum products thus has been escalating. The average of 2001-2006 gas price rised by 75% compared to the average of 1996-2001 (Ruhl, 2007). Therefore, it is indeed imminent in finding new source of oil, which is renewable and inexhaustible.

1.2 USE OF VEGETABLE OILS IN PHARMACEUTICALS

Vegetable oil has many advantages over mineral oil. Vegetable oils and fats and their derivatives, are renewable, biodegradable and harmless to the environment. They have been widely used as excipients and active ingredients in pharmaceutical and

cosmeceutical preparations.

The major components of vegetable oils are triglycerides. Triglycerides are esters of glycerol with three fatty acid molecules. The minor components include fatty alcohol, phytosterols, vitamins and phospholipids.

The oil uses depend mainly on the type and proportion of fatty acids which vary according to the species of plants. The oils with high content of unsaturated fatty acids (oleic, linoleic and linolenic acid) such as almond oil, apricot oil, avocado oil, soybean oil, cocoa oil and palm oil have emollient properties and are used in bar soaps, shampoos and hair conditioners, creams and lotions, lip balms and suntan products. However, the oils with high content of polyunsaturated fatty acid such as walnut oil and safflower oil are less stable and therefore less suitable to be used in its original state into the formulation. Safflower oil which contains 75-80% of linoleic acid need treatment to convert linoleic acid to oleic acid, to produce stable formulations.

Highly saturated vegetable oils are also gaining interest in the cosmetic and pharmaceutical fields due to their fewer tendencies to produce free radical from the oxidation process (Alias and Julianto, 2005). For instance, coconut oil is suitable for the preparation of cosmetic due to its high content of saturated fatty acids (lauric, 50% and myristic, 20%) and minor content of caprylic acid that exhibited antifungal activity (Alvarez and Rodriguez, 2000).

Vegetable oils are used as excipients in the formulation of cosmetics and pharmaceuticals in the form of its oil or its derivatives. For example, soybean oil has been used as oil phase in the preparation of intravenous fat emulsion for total parenteral nutrition (Intralipid[®]). Derivatives of castor oil, polyethoxylated castor oil (Cremofor[®]), is used as emulsifier and solubiliser in pharmaceutical preparations

3

containing volatile oils and fat-soluble vitamins and hydrogenated castor oil is used in some oral dosage forms and topical creams (Alvarez and Rodriguez, 2000).

Soybean oil is the major oil exploited in the parenteral preparation. Diprivan[®] (propofol), a general anaesthesia; Intralipid[®], a parenteral nutrition; Limethasone[®] (dexamethasone palmitate), for chronic rheumatoid arthritis and Lipo-NSAID[®] (flubiprofen axetil), for post-operative and cancer pain. These are commercially available emulsion formulations using soybean oil as the oil base.

1.2.1 Palm oil as an alternative

Palm oil is a form of edible vegetable oil obtained from the fruit of Elaeis Guineensis. Palm oil is the major commodity of Malaysia. Malaysia supplied 47% of world palm oil making it the largest world supplier in year 2005. It is now the first widely produced edible oil contributing to about 33% of world vegetable oil demand in 2005. Soybean oil only contributed 31% for the same year (Sumathi, Chai and Mohamed, 2007).

90% of the palm oil production is used as food and only 10% in non-food industry. The major uses in the foods are as cooking oil, shortening, margarine and spread and substitutes of dairy product. Non-food application of palm oil is mainly as a feedstock for manufacturing of soaps and oleochemicals. Palm oil based soaps produce high quality soap with better perfume retention and well accepted by all religions. The derivatives of oleochemicals are fatty acids, fatty acid methyl esters, fatty amines, fatty alcohols and glycerine. These compounds are widely used as essential ingredients in food emulsifiers, cosmetics, pharmaceuticals and fabric softeners. Palm oil has also been used in vulcanisation of natural rubber, natural rubber dipped-processing, lubricants, fertiliser industries and antistatic agent for