COPYRIGHT<sup>©</sup> INTERNATIONAL ISLAMIC UNIVERSITY MALAYSIA

## FABRICATION AND CHARACTERIZATION OF ZnO THIN FILM MEMRISTOR USING ULTRA DILUTE ELECTRODEPOSITION METHOD

BY

# FATIN BAZILAH FAUZI

A thesis submitted in fulfilment of the requirement for the degree of Master of Science (Materials Engineering)

Kuliyyah of Engineering International Islamic University Malaysia

FEBRUARY 2016

#### ABSTRACT

Memristor has become one of the alternatives to replace the current memory technologies. Instead of titanium dioxide (TiO<sub>2</sub>), many researches have been done to explore the compatibility of others transition metal oxide (TMO) by using various deposition methods. Recently, the compatibility of zinc oxide (ZnO) to be used as the active layer of memristor has been widely explored. Meanwhile, the usage of organic materials in electronic device has shown a rapid growth as the size demand of devices is increasingly smaller and faster. Future electronics industry depends on the development of organic base semiconductor devices due to their advantages. In this metal-insulator-metal of Au/ZnO-Cu<sub>2</sub>O-CuO/Cu study. the (MIM) and Au/ZnO/ITO/PET memristor were fabricated using dilute electrodeposition of zinc (Zn) and subsequent thermal oxidation methods at 773 K and 423 K respectively. The 15 s deposition gives the thinnest thin film, 80.67 nm for ZnO-Cu<sub>2</sub>O-CuO on Cu and 68.10 nm for ZnO on ITO coated PET. The deposited thin film was characterized via X-ray diffraction (XRD) and field emission scanning electron microscopy (FESEM). On Cu substrate, the XRD result indicates that Zn was oxidized to ZnO and has a wurzite structure. Meanwhile, Cu substrate also was oxidized to Cu<sub>2</sub>O and CuO. There was formation of needle like structure observed through FESEM after thermal oxidation method. While on ITO coated PET substrate, Zn was oxidized to wurzite ZnO as shown in XRD result with nodule structure of ZnO after the thermal oxidation method. Both Au/ZnO-Cu<sub>2</sub>O-CuO/Cu and Au/ZnO/ITO/PET sandwich memristive behavior were identified by the pinched hysteresis loop obtained from the I-V measurement. The high resistance state, HRS over low resistance state, LRS ratio of 1.110 and 1.067 respectively were obtained. Empirical study on thermodynamics of ZnO, Cu<sub>2</sub>O, CuO and diffusivity of  $Zn^{2+}$  and  $O^{2-}$  in ZnO shows that zinc vacancy was formed in ZnO layer, thus giving rise to its memristive behavior. The synthesized Au/ZnO-Cu<sub>2</sub>O-CuO/Cu and Au/ZnO/ITO/PET memristor show potential application in the production of a non-complex and low cost memristor. A flexible Au/ZnO/ITO coated PET memristor produces a comparable result to the Au/ZnO-Cu<sub>2</sub>O-CuO/Cu memristor and other previous studies on memristor. The flexible memristor is applicable to be fabricated using dilute electrodeposition at room temperature with low thermal oxidation process.

## ملخص البحث

أصبح memristor أحد البدائل لتكنولوجيات الذاكرة الحالية. تم دراسة تصنيع memristor ثاني أكسيد التيتانيوم (TiO2) باستخدام مختلف أساليب الترسيب. إضافة إلى ذلك، في الاونة الأخيرة تم القيام بمزيد من الأبحاث لاستكشاف مدى توافق أكسيد المعدن الانتقالي (TMO) مثل أكسيد الزنك (ZnO) لاستخدامه كطبقة نشطة لmemristor. وفي الوقت نفسه أظهراستخدام المواد العضوية في الأجهزة الإلكترونية نموا سريعا، حيث أن حجم الطلب على الأجهزة متزايد بشكل أصغر وسرعة اعلى. إن صناعة الالكترونيات في المستقبل تعتمد على تطوير أجهزة الموصلات الجزئية ذات القاعدة العضوية نظرا لمزاياها. في هذه الدراسة، تم تصنيع المعدن عازل المعدنية باستخدام الترسيب الكهربائي ل-Au/ZnO Cu2O-CuO/Cuحفف من الزنك (Zn) وAu/ZnO/ITO/PET memristor مخفف من الزنك (Zn) وطرق الأكسدة الحرارية اللاحقة. الترسيب ل 15 ثانية يعطي أنحف طبقة رقيقة و هي 80.67 نانومتر لZnO-Cu<sub>2</sub>O-Cu<sub>0</sub> و 68.10 نانومتر لZnO/ITO/PET . تم تشخيص الطبقة الرقيقة المرسبة من خلال الأشعة السينية (XRD) و الجحهر الإلكتروني ذات حقل انبعاث الماسح الضوئي (FESEM). أظهرت نتائج XRD أن الزنك Zn أكسد لأكسيد الزنك ZnO و لديه تركيب wurzite. وفي الوقت نفسه، تم أكسدة ركيزة النحاس أيضا إلى Cu<sub>2</sub>O وCu<sub>2</sub>O. لقد لوحظ تشكيل إبرة مثل أكسيد الزنك ZnO خلال الأكسدة الحرارية عند درجة حرارة هي K773. تم تحديد سلوك شطيرة memristive لكل من -K773. تم CuO/Cu و Au/ZnO/ITO/PET بواسطة حلقة التباطؤ الضيقة التي تم الحصول عليها عبر قياس I-V. نسبة التحويل المقاوم هو 1.110 و 1.067 على التوالي. تظهر الدراسة التجريبية على الديناميكا الحرارية لأكسيد الزنك Cu<sub>2</sub>O، ZnO و CuO ، والانتشارية ل+2n و O<sup>2-</sup> في ZnO أن فراغ الزنك تشكل في طبقة أكسيد الزنك ZnO، مما أدى إلى سلوكها الmemristive. يبين توليف Au/ZnO- memristor Cu<sub>2</sub>O-CuO/Cu و Au/ZnO/ITO/PET إمكانية التطبيق في إنتاج memristor بشكل غير معقد ومنخفض التكلفة. ينتج PET memristor المرن و المطلى بAu/ZnO/ITO نتائج مماثلة ل Au/ZnO-Cu<sub>2</sub>O-CuO/Cu memristor ومماثل لدراسات أخرى سابقة على memristor. يمكن تصنيع memristor المرن باستخدام الترسيب الكهربائي المخفف في درجة حرارة الغرفة و مع عملية أكسدة حرارية منخفضة.

### **APPROVAL PAGE**

I certify that I have supervised and read this study and that in my opinion, it conforms to acceptable standards of scholarly presentation and is fully adequate, in scope and quality, as a thesis for the degree of Master of Science (Materials Engineering).

Mohd Hanafi Ani Supervisor

.....

Iskandar Idris Yaacob Co-Supervisor

I certify that I have read this study and that in my opinion it conforms to acceptable standards of scholarly presentation and is fully adequate, in scope and quality, as a thesis for the degree of Master of Science (Materials Engineering).

Yose Fachmi Buys
Internal Examiner

Abdul Malik Marwan Ali External Examiner

This thesis was submitted to the Department of Manufacturing and Materials Engineering and is accepted as a fulfilment of the requirement for the degree of Master of Science (Materials Engineering).

> Mohammad Yeakub Ali Head, Department of Manufacturing and Materials Engineering

This thesis was submitted to the Kulliyyah of Engineering and is accepted as a fulfilment of the requirement for the degree of Master of Science (Materials Engineering).

Md. Noor Salleh

Dean, Kulliyyah of Engineering

### DECLARATION

I hereby declare that this thesis is the result of my own investigations, except where otherwise stated. I also declare that it has not been previously or concurrently submitted as a whole for any other degrees at IIUM or other institutions.

Fatin Bazilah Fauzi

Signature ..... Date .....

### INTERNATIONAL ISLAMIC UNIVERSITY MALAYSIA

### DECLARATION OF COPYRIGHT AND AFFIRMATION OF FAIR USE OF UNPUBLISHED RESEARCH

### FABRICATION AND CHARACTERIZATION OF ZnO THIN FILM MEMRISTOR USING ULTRA DILUTE ELECTRODEPOSITION METHOD

I declare that the copyright holder of this thesis is owned by Fatin Bazilah Fauzi.

Copyright © 2016 International Islamic University Malaysia. All rights reserved.

No part of this unpublished research may be reproduced, stored in a retrieval system, or transmitted, in any form or by any means, electronics, mechanical, photocopying, recording or otherwise without prior written permission of the copyright holder except as provided below

- 1. Any material contained in or derived from this unpublished research may only be used by others in their writing with due acknowledgement.
- 2. IIUM or its library will have the right to make and transmit copies (print or electronic) for institutional and academic purpose.
- 3. The IIUM library will have the right to make, store in a retrieval system and supply copies of this unpublished research if requested by other universities and research libraries.

By signing this form, I acknowledged that I have read and understand the IIUM Intellectual Property Right and Commercialization policy.

Affirmed by Fatin Bazilah Fauzi.

Signature

Date

### **ACKNOWLEDGEMENTS**

#### In the name of Allah, Most Gracious, Most Merciful. May the blessing and peace of Allah be upon our prophet Sayyidina Muhammad ibn Abdullah (peace be upon him), and upon his families and upon his companions and upon all his godly followers.

Alhamdulillah and syukur to Allah SWT for being able to finish my study. First of all, I would like to give my special thanks to my honorable supervisor, Dr. Mohd Hanafi Bin Ani for his countless guidance and advice throughout my study.

Also, I would like to express my gratitude to my colleagues (Mukhtaruddin Bin Musa, Edhuan Bin Ismail, Mohd Shukri Bin Sirat) and the members of Corrosion Lab 2 for their help and friendly support.

Many thanks to IIUM laboratory staffs in Kuliyyah of Engineering especially Br. Hairi (Corrosion Lab), Br. Sanadi (FESEM) and Br. Rahimie (XRD).

I would like also to thank the Ministry of Higher Education (MOHE) for Research Acculturation Grant Scheme (RAGS13-002-0065) and International Islamic University Malaysia for their funding and facilities.

The encouragement as a continued source of inspiration provided by my parents and family are fully appreciated.

## **TABLE OF CONTENTS**

Abstract in arabic	i i
Approval page	i
	,
Declaration page	
Copyright page	v
Acknowledgement	v
List of tables	Х
List of figures	х
List of abbreviations	X
List of symbols	X
CHAPTER 1: INTRODUCTION	
1.1 Limitations of current memory technologies	
1.2 The memristor	
1.2.1 The missing circuit element	
1.2.2 Memristor principle	
1.3 Problem statement	1
1.4 Objective of the research	1
1.5 Scope and limitation	1
1.6 Contribution of the research	1
1.7 Thesis organization	1
CHAPTER 2: LITERATURE REVIEW: THE MEMRISTOR	1
2.1 Introduction	1
2.2 Electrodeposition of memristors	1
2.3 Active layer materials in memristor	2
2.4 Previous studies on memristor.	2
2.5 Defect chemistry of ZnO	2
2.6 Chapter summary	3

CHAPTER 3: METHODOLOGY	36
3.1 Introduction.	36
3.2 Substrate cleaning	37
3.2.1 Copper substrate	38
3.2.2 ITO coated PET substrate	39
3.3 Fabrication of ZnO memristive device	39
3.3.1 Preparation of electrolytic bath	39
3.3.2 Electrodeposition process	40
3.3.3 Thermal oxidation process	43
3.3.3.1 Copper substrate	43
3.3.3.2 ITO/PET substrate	44
3.3.4 Gold coating	44
3.4 Characterization method.	44
3.4.1 Physical properties characterization	44
3.4.1.1 X-ray Diffraction (XRD)	44
3.4.1.2 Field emission scanning electron microscope	
(FESEM) and energy dispersion x-ray (EDX)	45
3.4.2 Electrical property characterization	45
3.5 Chapter summary	46
CHADTED 4. DECLIFTC AND ANALYCIC	10
4.1 Introduction	40 79
4.1 Introduction.	48
4.2 Experimental results and discussion	48
4.2.1 Cu substrate	40
4.2.1.1 X-ray diffraction (XKD) $\dots$	48
4.2.1.2 FESEM	52
4.2.1.5 Metal Oxide Oxidation thermodynamics	59
4.2.1.4 <i>I-V</i> measurements	02
4.2.1.5 Metal oxide growth's model and vacancy	67
defects formation	6/
4.2.2 ITO coated PET substrate	68
4.2.2.1 X-ray diffraction (XRD)	68 70
4.2.2.2 FESEM	70
4.2.2.3 Energy dispersion x-ray	78
4.2.2.4 <i>I-V</i> measurements	82
4.3 Chapter summary	8/

CHAPTER 5: CONCLUSION AND FUTURE DEVELOPMENT	89
5.1 Conclusion	89
5.2 Future development	91
REFERENCES	93
LIST OF PUBLICATIONS	101

## LIST OF TABLES

Table No.		Page No.
1.1	Advantages and disadvantages of memory system (Perez and De Rose, 2010; <i>Process Integration, Devices, and Structures</i> , 2011; Sunami, 2010)	4
1.2	Comparison between DRAM, RRAM (Memristor) and PCRAM	5
2.1	Summary of previous works on memristor	25
3.1	Zn electrodeposition times for 0.005 M electrolyte bath	43
3.2	List of materials	47
4.1	Average thickness of deposited ZnO-Cu <sub>2</sub> O-CuO thin film	53
4.2	Standard Gibbs reaction free energy change and its equilibrium oxygen partial pressure value calculated for ZnO, $Cu_2O$ and CuO at 773 K	61
4.3	Average thickness of deposited ZnO thin film	71

## LIST OF FIGURES

<u>Figure No.</u>		Page No.
1.1	Memory cell size shrinkage at DRAM in volume production (Sunami, 2010)	2
1.2	Overall current trend of memory system technology	3
1.3	The fourth missing memristive systems as relation between electric charges and magnetic flux in an electric circuit (Strukov et al., 2008) (a) and Ideal <i>I-V</i> curves of each circuit element (Williams, 2008) (b)	8
1.4	Memristor device in an equivalent circuit (Strukov et al., 2008)	10
1.5	Pinched hysteresis loop of memristor predicted model by Chua (1971) (a) and Schematic $I-V$ hysteretic for bipolar switching (b)	12
1.6	Illustration on the movements of oxygen vacancies in memristor	14
1.7	Trends in device chip and feature size of CMOS device (Sunami, 2010)	15
2.1	The schematic diagram of Zn and ZnO interface during the ZnO growth	30
2.2	Four ions transportation mechanisms in oxidation reaction (Choopun et al., 2010)	32
2.3	Zinc diffusivity in ZnO from experiment and calculation (Erhart and Albe, 2006a)	33
2.4	Oxygen diffusivity in ZnO from experiment and calculation (Erhart and Albe, 2006b)	33
3.1	Overall process flow of methodology	37

3.2	Process flow of cleaning Cu substrate	38
3.3	Conductivity of aqueous ZnCl <sub>2</sub> at room temperature (Zhang, 1996)	40
3.4	Schematic diagram for electrolytic cell holder	41
3.5	Setup of the cell	41
3.6	Schematic diagram of electrodeposition process	42
3.7	Schematic diagram of the device stack	44
3.8	Schematic diagram of <i>I-V</i> measurement for Au/ZnO/ITO/PET	46
4.1	XRD pattern on Zn/Cu before the thermal oxidation	49
4.2	XRD pattern on ZnO-Cu <sub>2</sub> O-CuO/Cu after the thermal oxidation	50
4.3	XRD pattern on bare Cu substrate and after the thermal oxidation	51
4.4	Field emission scanning electron microscope (FESEM) cross- sectional of ZnO-Cu <sub>2</sub> O-CuO/Cu junction of deposition time at 15 s (a), 30 s (b), 60 s (c) and 120 s (d)	53
4.5	Effect of deposition time on $ZnO-Cu_2O-CuO$ thin film thickness	55
4.6	$D^2$ against <i>t</i> curve following the parabolic rate law on ZnO-Cu <sub>2</sub> O-CuO thin film	55
4.7	Field emission scanning electron microscope (FESEM) with low 20k (i) and high 40k (ii) magnification of surface morphology of Zn before the thermal oxidation of deposition time at 15 s (a), 30 s (b), 60 s (c) and 120 s (d)	57
4.8	Field emission scanning electron microscope (FESEM) with low 20k (i) and high 60k (ii) magnification of surface morphology of ZnO-Cu <sub>2</sub> O-CuO after the thermal oxidation of deposition time at 15 s (a), 30 s (b), 60 s (c) and 120 s (d)	58

4.9	Metal oxide formation model based on the calculated oxygen partial pressure for ZnO, Cu <sub>2</sub> O and CuO	61
4.10	<i>I-V</i> hysteresis loops from synthesized ZnO thin films at 15 s (a), 30 s (b), 60 s (c) and 120 s (d)	62
4.11	Difference of HRS and LRS (a) and Switching ratio (HRS/LRS) (b) of deposition time at 15 s, 30 s, 60 s and 120 s	64
4.12	Semi log scale of <i>I-V</i> curves in 100 cycles for 15 s deposited Au/ZnO-Cu <sub>2</sub> O-CuO/Cu. Inset: semi log scale of <i>I-V</i> curves with arrow of sweep direction	65
4.13	Resistance cycle characteristics of 15 s Au/ZnO-Cu <sub>2</sub> O-CuO/Cu measured at 0.25 V for 100 Cycles	66
4.14	Schematic diagram of ions transportation in thermal oxidation reaction	67
4.15	XRD pattern of Zn/ITO/PET before the thermal oxidation	69
4.16	XRD pattern of ZnO/ITO/PET after the thermal oxidation	69
4.17	Field emission scanning electron microscope (FESEM) cross- sectional of ZnO/ITO Coated PET junction of deposition time at 15 s (a), 30 s (b), 60 s (c) and 120 s (d)	70
4.18	The effect of deposition time on ZnO thin film thickness	72
4.19	$D^2$ against <i>t</i> curve following the parabolic rate law on ZnO thin film	72
4.20	<i>I</i> against <i>t</i> graph at 2 V for Zn/Cu (a) and Zn/ITO/PET (b)	74
4.21	Field emission scanning electron microscope (FESEM) with low 20k (i) and high 40k (ii) magnification of surface morphology of Zn before the thermal oxidation of deposition time at 15 s (a), 30 s (b), 60 s (c) and 120 s (d)	76

4.22	Field emission scanning electron microscope (FESEM) with low 20k (i) and high 40k (ii) magnification of surface morphology of ZnO after the thermal oxidation of deposition time at 15 s (a), 30 s (b), 60 s (c) and 120 s (d)	77
4.23	EDX pattern on surface of ZnO/ITO/PET	78
4.24	EDX pattern on cross section of ZnO/ITO/PET	79
4.25	EDX mapping elements distribution of ZnO/ITO coated PET	81
4.26	<i>I-V</i> hysteresis loops from synthesized ZnO thin films at 15 s (a), 30 s (b), 60 s (c) and 120 s (d)	82
4.27	Difference of HRS and LRS (a) and Switching ratio (HRS/LRS) (b) of deposition time at 15 s, 30 s, 60 s and 120 s	84
4.28	Semi log scale of <i>I-V</i> curves in 100 cycles for 15 s deposited Au/ZnO/ITO/PET	85
4.29	Resistance cycle characteristics of 15 s Au/ZnO/ITO/PET measured at 3.0 V for 100 cycles	86
4.30	Comparison of HRS/LRS ratio of this study between Au/ZnO-Cu <sub>2</sub> O-CuO/Cu and Au/ZnO/ITO/PET memristors with previous works (Han et al., 2011, Jia et al., 2012, Kumar et al., 2012)	87

## LIST OF ABBREVIATIONS

1-D	1-Dimensional
1T-1M	1Transistor- 1Memristor
A.u	Arbitrary unit
Ag/ AgCl	Silver chloride
Al	Aluminium
Al <sub>2</sub> O <sub>3</sub>	Aluminium (III) oxide
ALD	Atomic layer deposition
Au	Aurum
С	Carbon
CMOS	Complimentary metal-oxide-semiconductor
Cu	Copper
Cu <sub>2</sub> O	Copper (I) oxide
CuO	Copper (II) oxide
CVD	Chemical vapor deposition
DC	Direct current
DRAM	Dynamic random access memory
EDX	Energy dispersion x-ray
FESEM	Field emission-scanning electron microscope
FTO	Fluorine doped tin oxide
FWHM	Full width at half width maxima
GO	Graphene oxide
HP	Hewlett-Packard

hr	Hour
HRS	High resistance state (same as $R_{OFF}$ )
In	Indium
ITO	Indium tin oxide
I-V	Current-voltage
JCPDS	Joint committee on powder diffraction standards
LRS	Low resistance state (same as $R_{ON}$ )
М	Molar
MRAM	Magnetoresistive RAM
MgO	Magnesium oxide
MIM	Metal-insulator-metal
min	Minute
n	Refractive index
N <sub>2</sub>	Nitrogen
NAND	Not AND
Nb <sub>2</sub> O <sub>5</sub>	Niobium oxide
NBE	Near band edge
Ni	Nickel
NiO	Nickel (II) oxide
O <sub>2</sub>	Oxygen
PCRAM	Phase change RAM
PET	Polyethylene terephthalate
PF	Poole-Frenkel
PL	Photoluminescence
PLD	Pulsed laser deposition
Pt	Platinum

RAM	Random access memory
RF	Radio frequency
RRAM	Resistive RAM
RT	Room temperature
RTP	Rapid thermal process
SCCM	Standard cubic centimeters per minute
SEM	Scanning electron microscope
Si	Silicon
SiO <sub>2</sub>	Silicon dioxide
SRAM	Static random access memory
SrZrO <sub>3</sub>	Strontium zirconate
Ti	Titanium
TiO <sub>2</sub>	Titanium dioxide
ТМО	Transition metal oxide
US	United States
LIV/	
UV	Ultraviolet
w	Ultraviolet Tungsten
W XRD	Ultraviolet Tungsten X-ray diffraction
W XRD Zn	Ultraviolet Tungsten X-ray diffraction Zinc
W XRD Zn ZnCl <sub>2</sub>	Ultraviolet Tungsten X-ray diffraction Zinc Zinc chloride
W XRD Zn ZnCl <sub>2</sub> ZnO	Ultraviolet Tungsten X-ray diffraction Zinc Zinc chloride Zinc oxide

## LIST OF SYMBOLS

$\Delta G$	Gibbs reaction free energy change
$\Delta G^{\circ}$	Standard Gibbs reaction free energy change
θ	Theta
λ	Wavelength
ε <sub>r</sub>	Optical dielectric constant
Ω	Ohm
k	Reaction constant
$arphi_m$	Magnetic flux
$\mu m^2$	Square micrometer
$\mu_{v}$	Dopant mobility
А	Ampere
Å	Angstrom
atm	Standard atmospheric pressure
С	Concentration of oxygen
$C_s$	Storage capacitance
D	Diffusion coefficient (diffusivity)
d	Lattice distance
D	Thin film thickness
eV	Electron volt
fF	Femtofarads
G	Giga
I	Current

J	Diffusive flux
Κ	Kelvin
k	Kilo
k <sub>p</sub>	Parabolic rate constant
L	Inductance
ł	Liquid
М	Memristance
т	Slope of straight line
mm <sup>2</sup>	Square millimeter
ml	Milliliter
nm	Nanometer
Р	Power
Pa	Pascal
$P_g$	Gas pressure
<i>P</i> <sub>02</sub>	Oxygen partial pressure
q	Charge
R	Ideal gas constant
R <sub>OFF</sub>	Off-state resistance (same as HRS)
RON	On-state resistance (same as LRS)
S	Seconds
S	Solid
sq	Square
Т	Temperature
t	Time
V	Voltage

V Volts

*x* Displacement (position of length of diffusion)

#### **CHAPTER 1**

#### **INTRODUCTION**

#### **1.1 LIMITATIONS OF CURRENT MEMORY TECHNOLOGIES**

The reliance of people towards technology drove the development of computers and other technology devices. Innovation of smaller electronic devices with better performance and capacity give a huge impact to the memory storage system. As the technology developed, it can be seen that the size of the devices also are getting smaller. The smaller the device is, the more efficient it will be, but this brings forth the challenge in increasing the physical memory capacity. Presently used memory technologies such as dynamic random access memory (DRAM), static random access memory (SRAM), and NAND flash are facing design challenge due to the continued scale down in physical size.

The memory system is the most crucial component in any electronic devices. It is where the computer keeps its current programs and data that are in use. With the advances in the technology of electronic devices, the memory technology has met its maximum limit to keep on par with the demand for smaller size and higher capacity memory storage.

Figure 1.1 shows the trends in chip size, memory cell size and storage capacitance in response to DRAM generation by (Sunami, 2010). The increase of bit size of DRAM by a factor of 10<sup>6</sup> from 1 kbit to 1 Gbit caused the enlargement of the chip size up to 10 times. As mentioned previously, as the fabrication of increasingly smaller memory cell size grows harder, it could be expected to meet a limit on producing small chip sized with bigger DRAM size.

1



Figure 1.1 Memory cell size shrinkage at DRAM in volume production (Sunami, 2010)

On the other hand, the manufacturing cost of DRAM elevated up due to the improvement route in size and capacity to achieve the required specifications. Therefore, various development efforts have been focusing on reduction of manufacturing cost. Stated by Kwon in his thesis, worldwide DRAM industry has lost over 10 billion US Dollars during 2006 to 2008 because of the technological innovations and DRAM scaling. The cell size of DRAM has to be reduced every year by 30 % to satisfy the latest market demand while the price per bit of DRAM drops 26 % annually. This will end up in a big loss for the DRAM manufacturer (Kwon, 2013). A significant portion of the total system power and the total system cost is spent in the memory system with the increasing size of the memory system (Qureshi, Srinivasan, and Rivers, 2009). Figure 1.2 shows the overall current trend of the memory system where the target size approaches 10 nm, with higher capacity and

speed. But as to reduce the size and improve the performance of the device, the manufacturing cost increases because of the increase in power consumption by the device itself and its cooling system itself as the structure of the device more complex.



Figure 1.2 Overall current trend of memory system technology

The memory system is one of the most critical components of modern computers. Dynamic random access memory (DRAM) and static random access memory (SRAM) are the main system memory used in any electronic devices. They can be considered as the crucial components of the memory system with fast response and good performance.

But, DRAM and SRAM are the volatile memory system which they cannot store the memory data without the power supply (Perez and De Rose, 2010; Lian, 2014). The current memory devices cost billions dollars to keep pace with the current trend and capacity. It has reaching their maximum physical limitation when demands of the smaller size memory devices arise.